

Magnetska svojstva metala

- U magnetskom polju sustav se magnetizira
- Ukupna magnetizacija u materijalu:

$$M = \frac{\text{Rezultantni dipolni moment}}{\text{Jedinični volumen}}$$

- Izotropni kristali: $\vec{M} = \chi \vec{H}$

- Susceptibilnost sustava - χ

PERIODNI SUSTAV ELEMENATA

| | | | | | | | | | | | | | | | | | |
|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|
| 1 H 1.008 | | | | | | | | | | | | | | | | | 2 He 4.003 |
| 3 Li 6.941 | 4 Be 9.012 | | | | | | | | | | | 5 B 10.81 | 6 C 12.01 | 7 N 14.01 | 8 O 16.00 | 9 F 19.00 | 10 Ne 20.18 |
| 11 Na 22.99 | 12 Mg 24.30 | ← VIII → | | | | | | | | | | 13 Al 26.98 | 14 Si 28.09 | 15 P 30.97 | 16 S 32.07 | 17 Cl 35.05 | 18 Ar 39.95 |
| 19 K 39.10 | 20 Ca 40.08 | 21 Sc 44.96 | 22 Ti 47.87 | 23 V 50.94 | 24 Cr 52.00 | 25 Mn 54.94 | 26 Fe 55.85 | 27 Co 58.93 | 28 Ni 58.69 | 29 Cu 63.55 | 30 Zn 65.39 | 31 Ga 69.72 | 32 Ge 72.61 | 33 As 74.92 | 34 Se 78.96 | 35 Br 79.90 | 36 Kr 83.80 |
| 37 Rb 85.47 | 38 Sr 87.62 | 39 Y 88.91 | 40 Zr 91.22 | 41 Nb 92.91 | 42 Mo 95.94 | 43 Tc 98.91 | 44 Ru 101.1 | 45 Rh 102.9 | 46 Pd 106.4 | 47 Ag 107.9 | 48 Cd 112.4 | 49 In 114.8 | 50 Sn 118.7 | 51 Sb 121.8 | 52 Te 127.6 | 53 I 126.9 | 54 Xe 131.3 |
| 55 Cs 132.9 | 56 Ba 137.3 | La-Lu | 72 Hf 178.5 | 73 Ta 180.9 | 74 W 183.8 | 75 Re 186.2 | 76 Os 190.2 | 77 Ir 192.2 | 78 Pt 195.1 | 79 Au 197.0 | 80 Hg 200.6 | 81 Tl 204.4 | 82 Pb 207.2 | 83 Bi 209.0 | 84 Po 210.0 | 85 At 210.0 | 86 Rn 222.0 |
| 87 Fr 223.0 | 88 Ra 226.0 | Ac-Lr | 104 Db | 105 Jl | 106 Rf | 107 Bh | 108 Hn | 109 Mt | 110 Uun | 111 Uuu | | | | | | | |



LANTANIDI

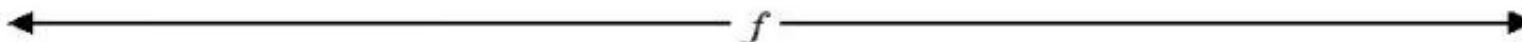
4f-orbitale

| | | | | | | | | | | | | | | |
|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|
| 57 La 138.9 | 58 Ce 140.1 | 59 Pr 140.9 | 60 Nd 144.2 | 61 Pm 146.9 | 62 Sm 150.4 | 63 Eu 152.0 | 64 Gd 157.2 | 65 Tb 158.9 | 66 Dy 162.5 | 67 Ho 164.9 | 68 Er 167.3 | 69 Tm 168.9 | 70 Yb 173.0 | 71 Lu 175.0 |
|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|

AKTINIDI

5f-orbitale

| | | | | | | | | | | | | | | |
|--------------------------|--------------------------|--------------------------|-------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|---------------------------|---------------------------|---------------------------|---------------------------|
| 89 Ac 227.0 | 90 Th 232.0 | 91 Pa 231.0 | 92 U 238.0 | 93 Np 237.0 | 94 Pu 239.1 | 95 Am 241.1 | 96 Cm 244.1 | 97 Bk 249.1 | 98 Cf 252.1 | 99 Es 252.1 | 100 Fm 257.1 | 101 Md 258.1 | 102 No 259.1 | 103 Lr 262.1 |
|--------------------------|--------------------------|--------------------------|-------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|---------------------------|---------------------------|---------------------------|---------------------------|



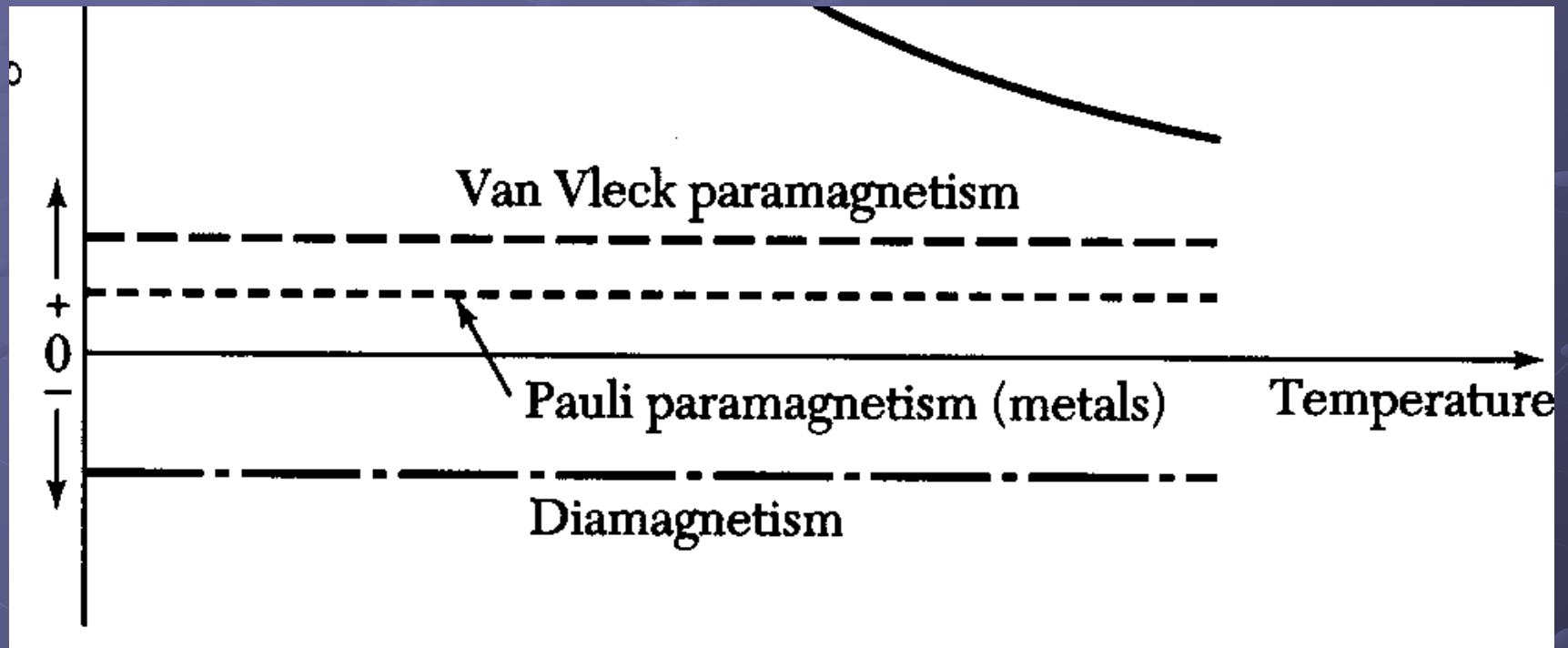
● Razlikujemo:

- Slabe magnete

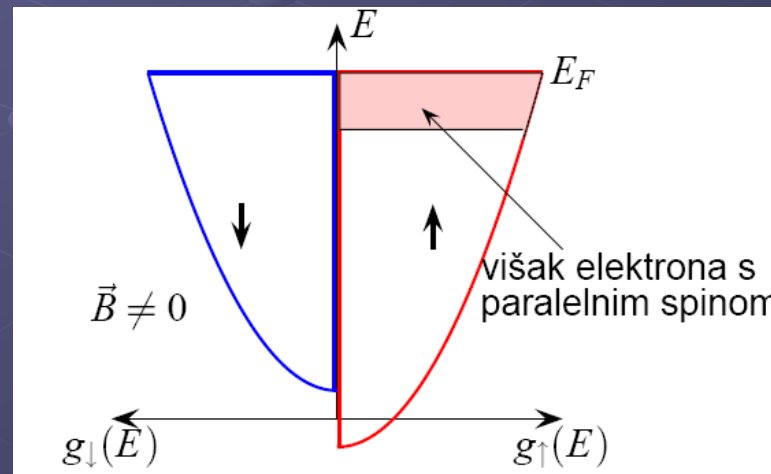
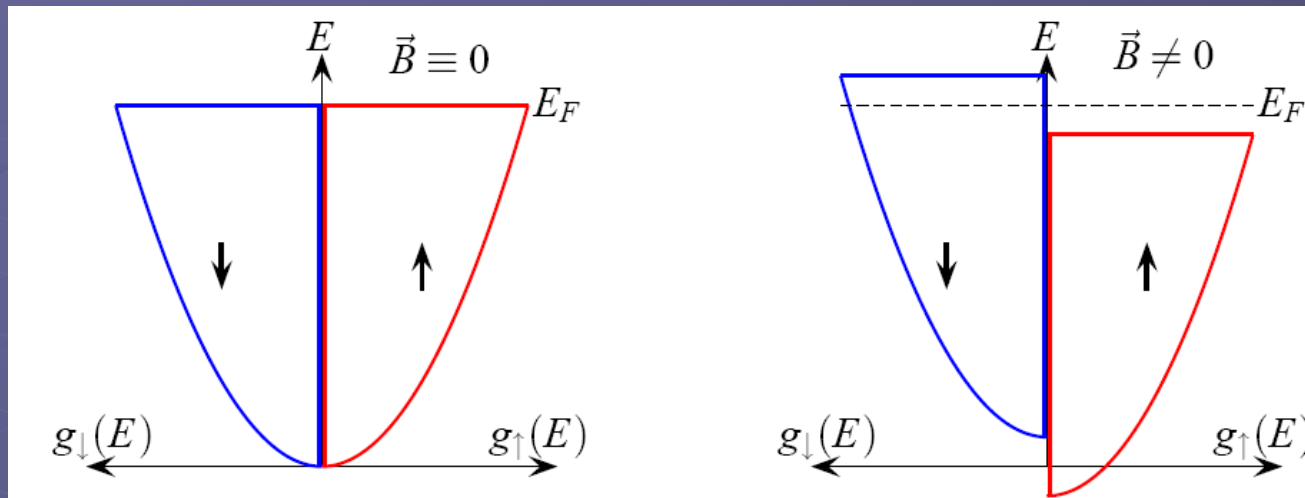
- Dijamagneti
- Paramagneti

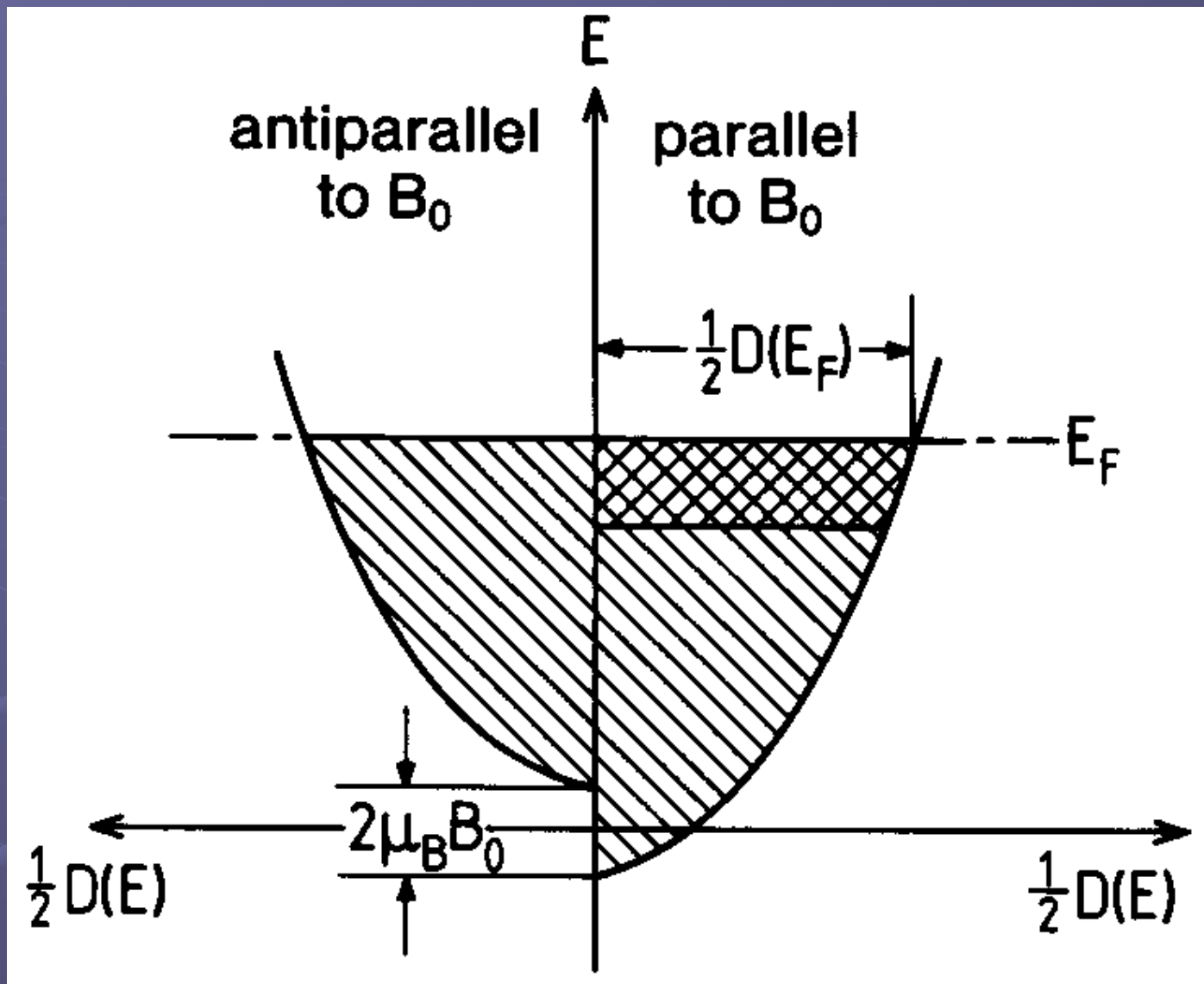
- Jake magnete

- Feromagneti
- Antiferomagneti
- Ferimagneti

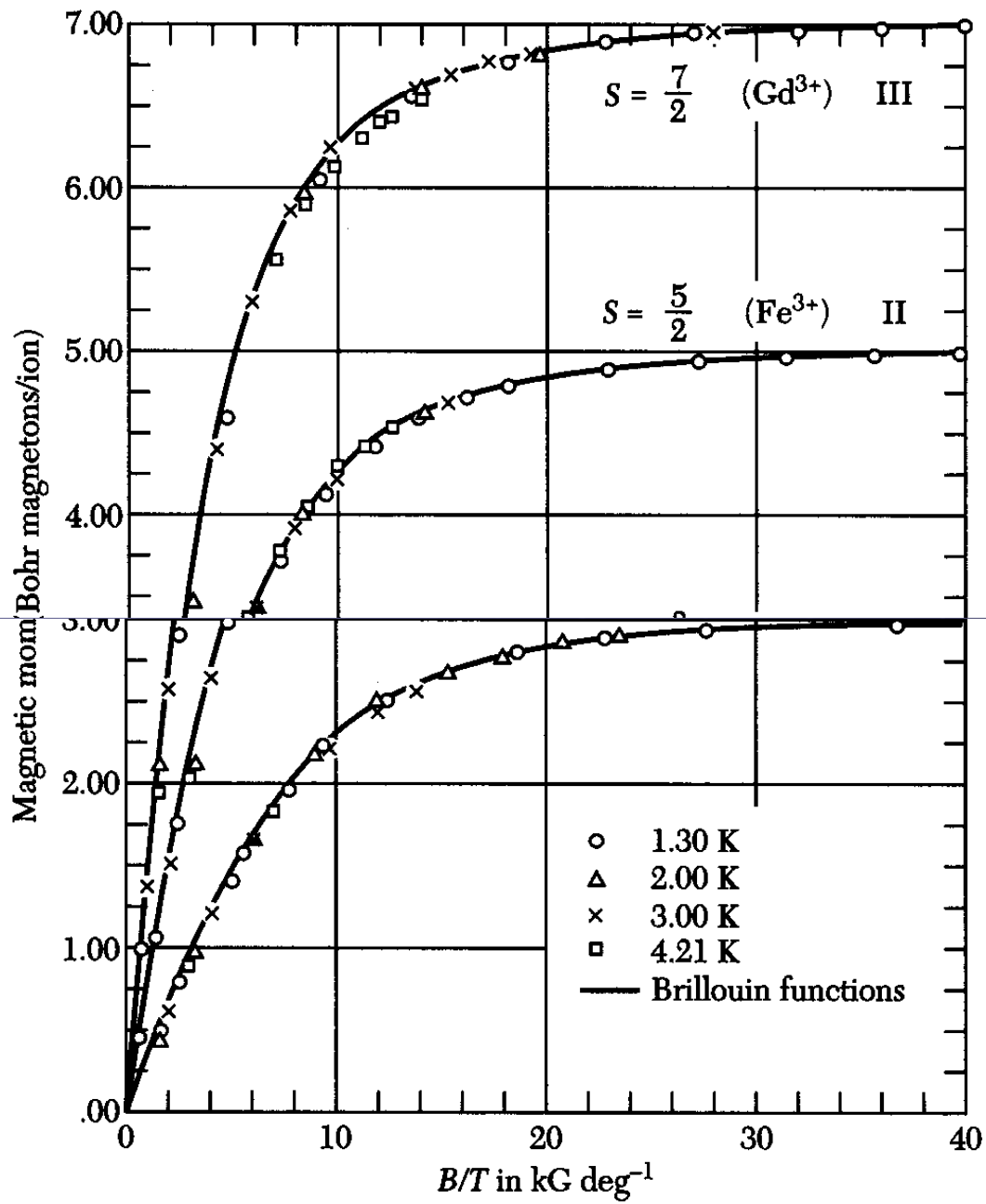


Paulijev paramagnetizam





| ELEMENT (AND IONIZATION) | BASIC ELECTRON CONFIGURATION | GROUND- STATE TERM | CALCULATED ^b p | | MEASURED ^c p |
|--------------------------------|------------------------------------|-------------------------------|-----------------------------|---------------------|---------------------------|
| | | | ($J = S$) | ($J = L \pm S $) | |
| Ti ³⁺ | 3d ¹ 4s ² | ² D _{3/2} | 1.73 | 1.55 | — |
| V ⁴⁺ | 3d ¹ | ² D _{3/2} | 1.73 | 1.55 | 1.8 |
| V ³⁺ | 3d ² | ³ F ₂ | 2.83 | 1.63 | 2.8 |
| V ²⁺ | 3d ³ | ⁴ F _{3/2} | 3.87 | 0.77 | 3.8 |
| Cr ³⁺ | 3d ³ | ⁴ F _{3/2} | 3.87 | 0.77 | 3.7 |
| Mn ³⁺ | 3d ⁵ | ⁶ S _{5/2} | 5.92 | 5.92 | 4.0 |
| Cr ²⁺ | 3d ⁴ | ⁵ D ₀ | 4.90 | 0 | 4.8 |
| Mn ³⁺ | 3d ⁴ | ⁵ D ₀ | 4.90 | 0 | 5.0 |
| Mn ²⁺ | 3d ⁵ | ⁶ S _{5/2} | 5.92 | 5.92 | 5.9 |
| Fe ³⁺ | 3d ⁵ | ⁶ S _{5/2} | 5.92 | 5.92 | 5.9 |
| Fe ²⁺ | 3d ⁶ | ⁵ D ₄ | 4.90 | 6.70 | 5.4 |
| Co ²⁺ | 3d ⁷ | ⁴ F _{9/2} | 3.87 | 6.54 | 4.8 |
| Ni ²⁺ | 3d ⁸ | ³ F ₄ | 2.83 | 5.59 | 3.2 |
| Cu ²⁺ | 3d ⁹ | ² D _{5/2} | 1.73 | 3.55 | 1.9 |

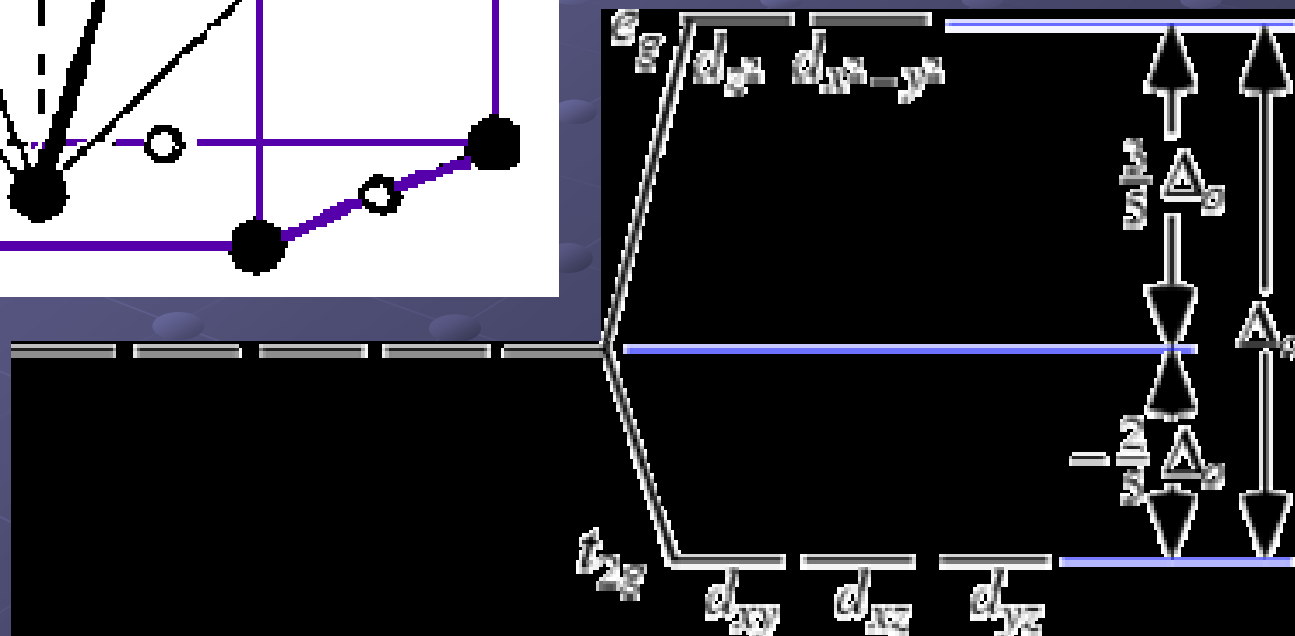
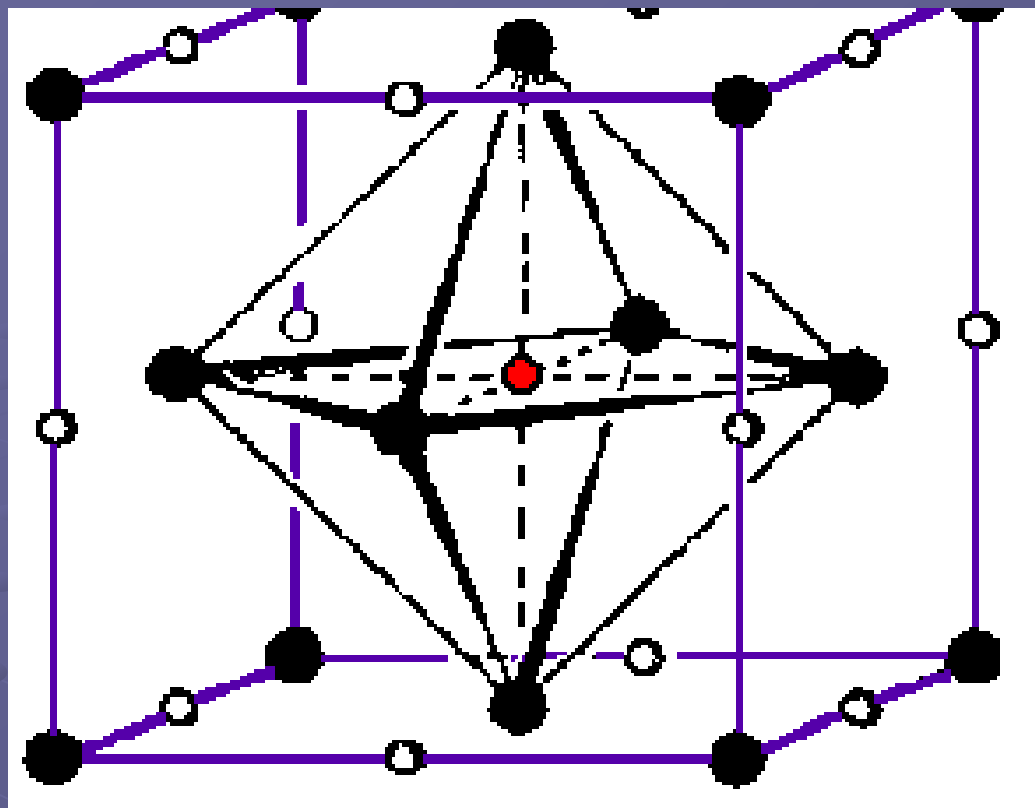


Hundova Pravila

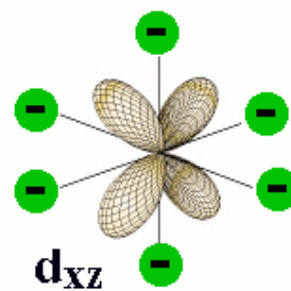
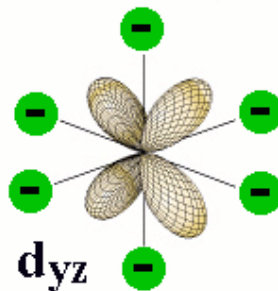
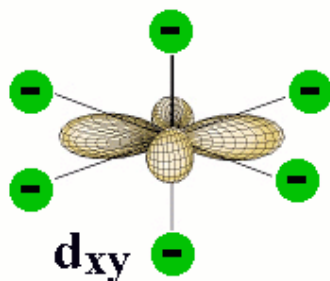
$J=L+S$

| d-shell ($l = 2$) | | | | | | | | | | | | |
|---------------------|------------|------|------|-------|-------|-------|--------------------|-----|--------------------|---------------|-----------------|----------------|
| n | $l_z = 2,$ | $1,$ | $0,$ | $-1,$ | -2 | S | $L = \Sigma l_z $ | J | | SYMBOL | | |
| 1 | ↓ | | | | | 1/2 | 2 | 3/2 | } $J = L - S $ | ${}^2D_{3/2}$ | | |
| 2 | ↓ | ↓ | | | | 1 | 3 | 2 | | 3F_2 | | |
| 3 | ↓ | ↓ | ↓ | | | 3/2 | 3 | 3/2 | | ${}^4F_{3/2}$ | | |
| 4 | ↓ | ↓ | ↓ | ↓ | | 2 | 2 | 0 | | 5D_0 | | |
| 5 | ↓ | ↓ | ↓ | ↓ | ↓ | 5/2 | 0 | 5/2 | | ${}^6S_{5/2}$ | | |
| 6 | ↑ | ↑ | ↑ | ↑ | ↑ | 2 | 2 | 4 | } $J = L + S$ | 5D_4 | | |
| 7 | ↑ | ↑ | ↑ | ↑ | ↑ | 3/2 | 3 | 9/2 | | ${}^4F_{9/2}$ | | |
| 8 | ↑ | ↑ | ↑ | ↑ | ↑ | 1 | 2 | 4 | | 3F_4 | | |
| 8 | ↑ | ↑ | ↑ | ↑ | ↑ | 1 | 2 | 4 | | 3F_4 | | |
| f-shell ($l = 3$) | | | | | | | | | | | | |
| n | $l_z = 3,$ | $2,$ | $1,$ | $0,$ | $-1,$ | $-2,$ | -3 | S | $L = \Sigma l_z $ | J | | |
| 1 | ↓ | | | | | | | 1/2 | 3 | 5/2 | } $J = L - S $ | |
| 2 | ↓ | ↓ | | | | | | 1 | 5 | 4 | | 3H_4 |
| 3 | ↓ | ↓ | ↓ | | | | | 3/2 | 6 | 9/2 | | ${}^4I_{9/2}$ |
| 4 | ↓ | ↓ | ↓ | ↓ | | | | 2 | 6 | 4 | | 5I_4 |
| 5 | ↓ | ↓ | ↓ | ↓ | ↓ | | | 5/2 | 5 | 5/2 | | ${}^6H_{5/2}$ |
| 6 | ↓ | ↓ | ↓ | ↓ | ↓ | ↓ | | 3 | 3 | 0 | } $J = L + S$ | |
| 7 | ↓ | ↓ | ↓ | ↓ | ↓ | ↓ | ↓ | 7/2 | 0 | 7/2 | | ${}^8S_{7/2}$ |
| 8 | ↑ | ↑ | ↑ | ↑ | ↑ | ↑ | ↑ | 3 | 3 | 6 | | 7F_6 |
| 9 | ↑ | ↑ | ↑ | ↑ | ↑ | ↑ | ↑ | 5/2 | 5 | 15/2 | | ${}^6H_{15/2}$ |
| 10 | ↑ | ↑ | ↑ | ↑ | ↑ | ↑ | ↑ | 2 | 6 | 8 | | 5I_8 |
| 11 | ↑ | ↑ | ↑ | ↑ | ↑ | ↑ | ↑ | 3/2 | 6 | 15/2 | | ${}^4I_{15/2}$ |
| 12 | ↑ | ↑ | ↑ | ↑ | ↑ | ↑ | ↑ | 1 | 5 | 6 | | 3H_6 |
| 13 | ↑ | ↑ | ↑ | ↑ | ↑ | ↑ | ↑ | 1/2 | 3 | 7/2 | | ${}^2F_{7/2}$ |
| 14 | ↑ | ↑ | ↑ | ↑ | ↑ | ↑ | ↑ | 0 | 0 | 0 | | 1S_0 |

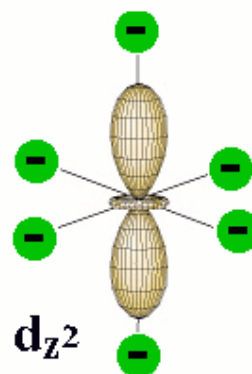
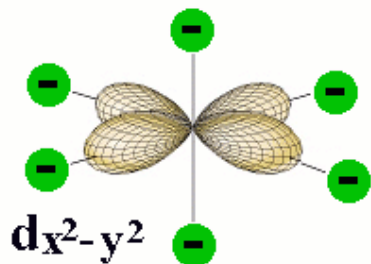
*↑ = spin 1/2; ↓ = spin -1/2.

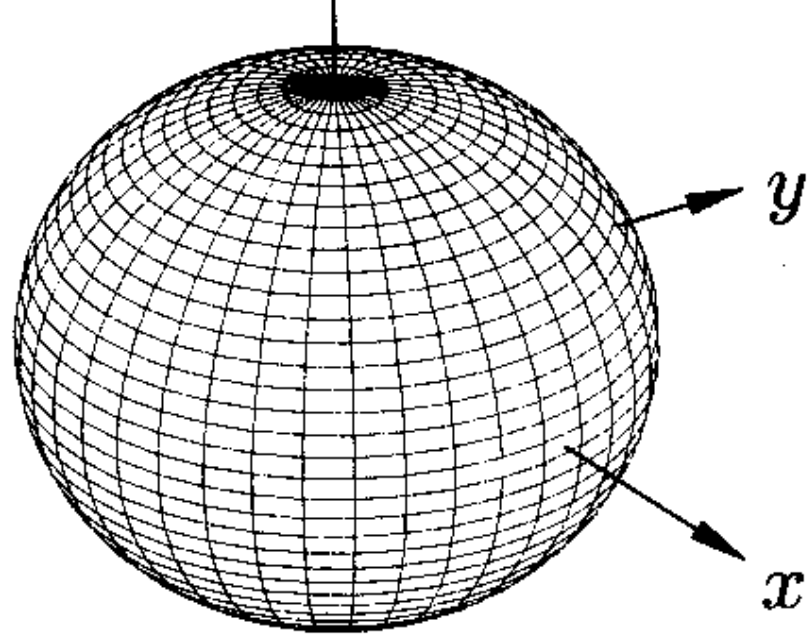


**Lower
Energy
Levels**

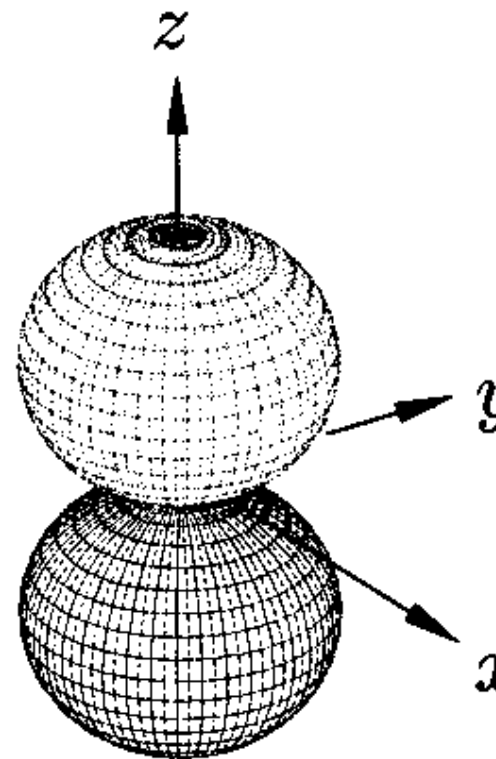
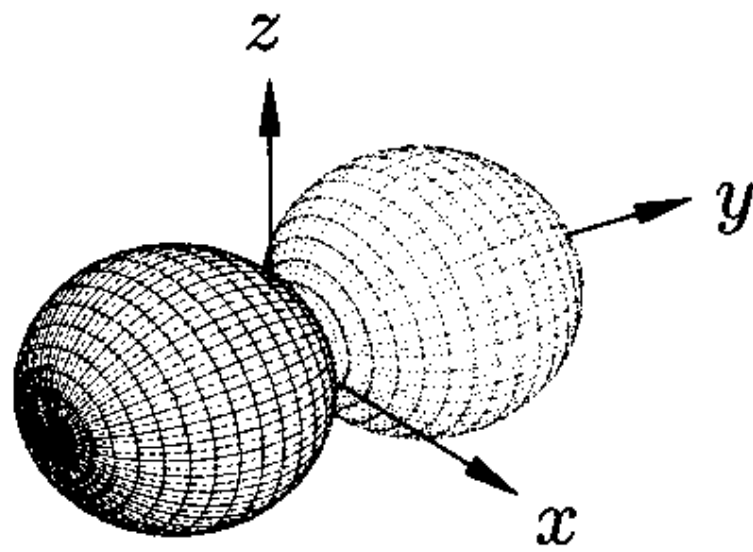
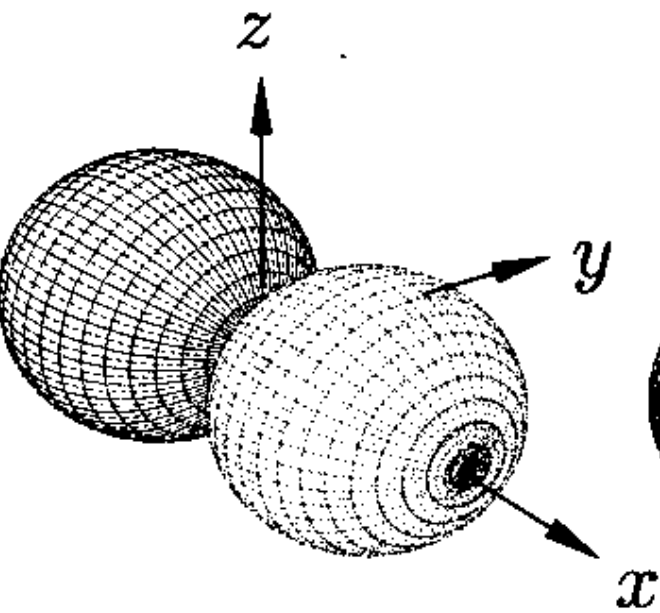


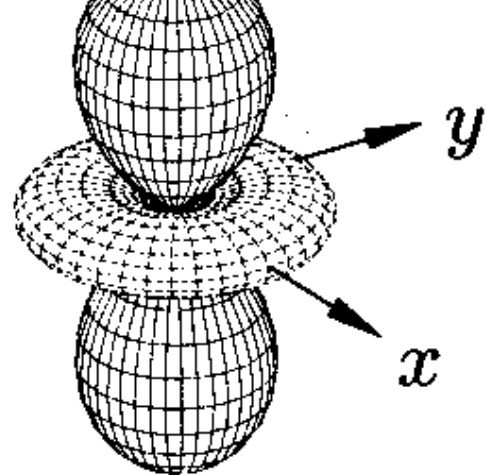
**Higher
Energy
Levels**



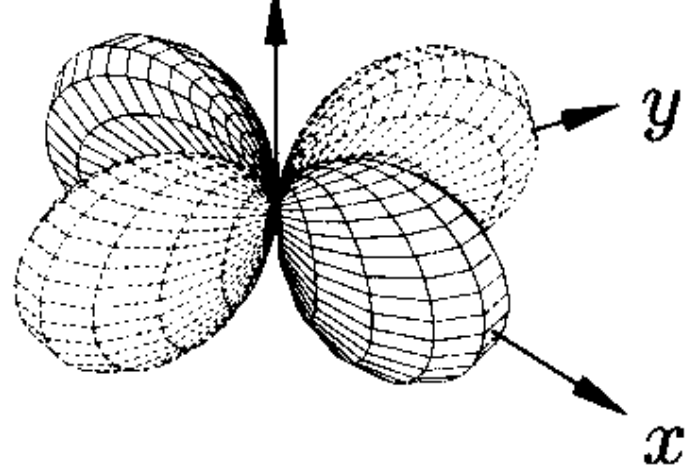


S

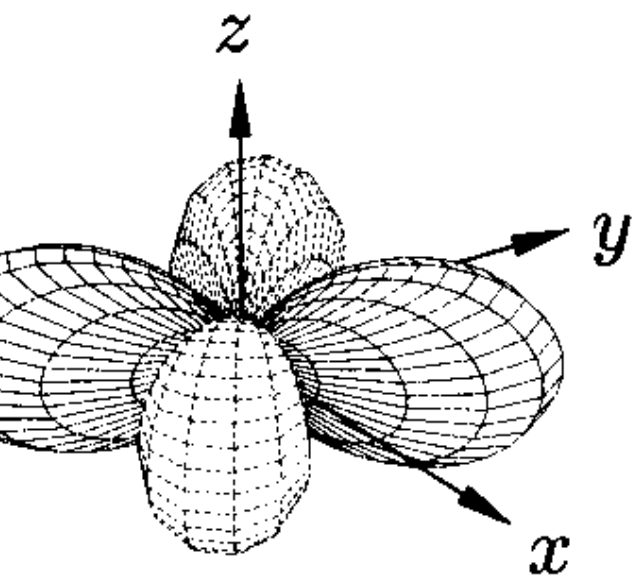




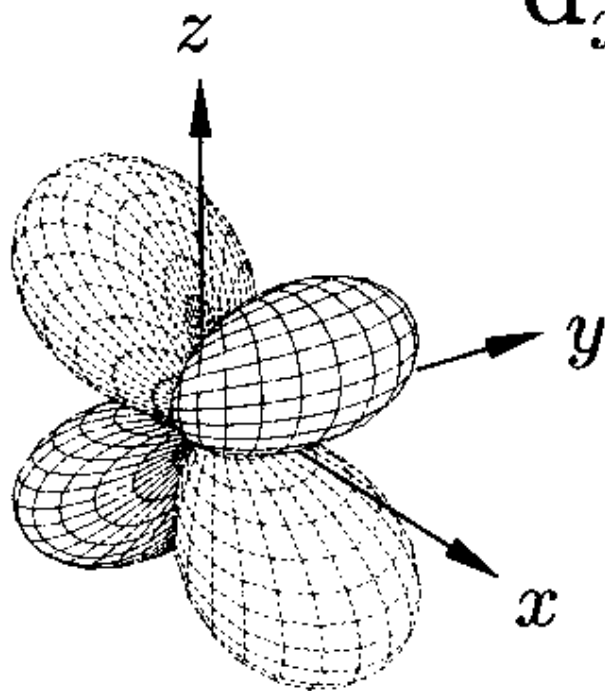
d_{z^2}



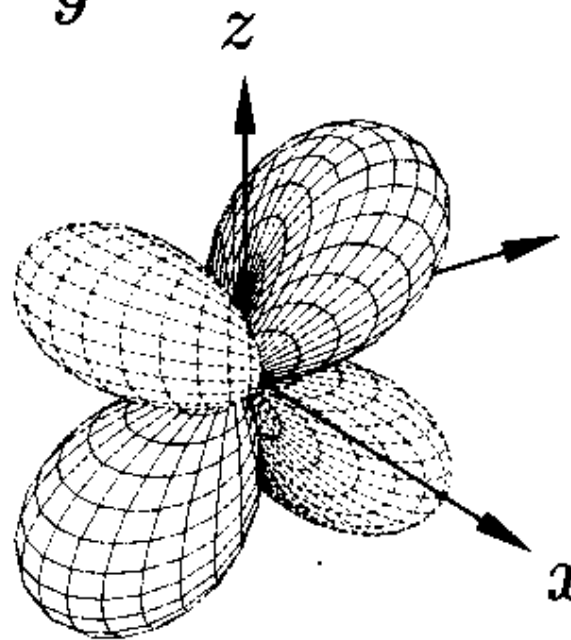
$d_{x^2-y^2}$



d_{xy}



d_{xz}



d_{yz}

● Za $\vec{H} \neq 0$

- $X < 0$ dij magnetizam
- $X > 0$ paramagnetizam

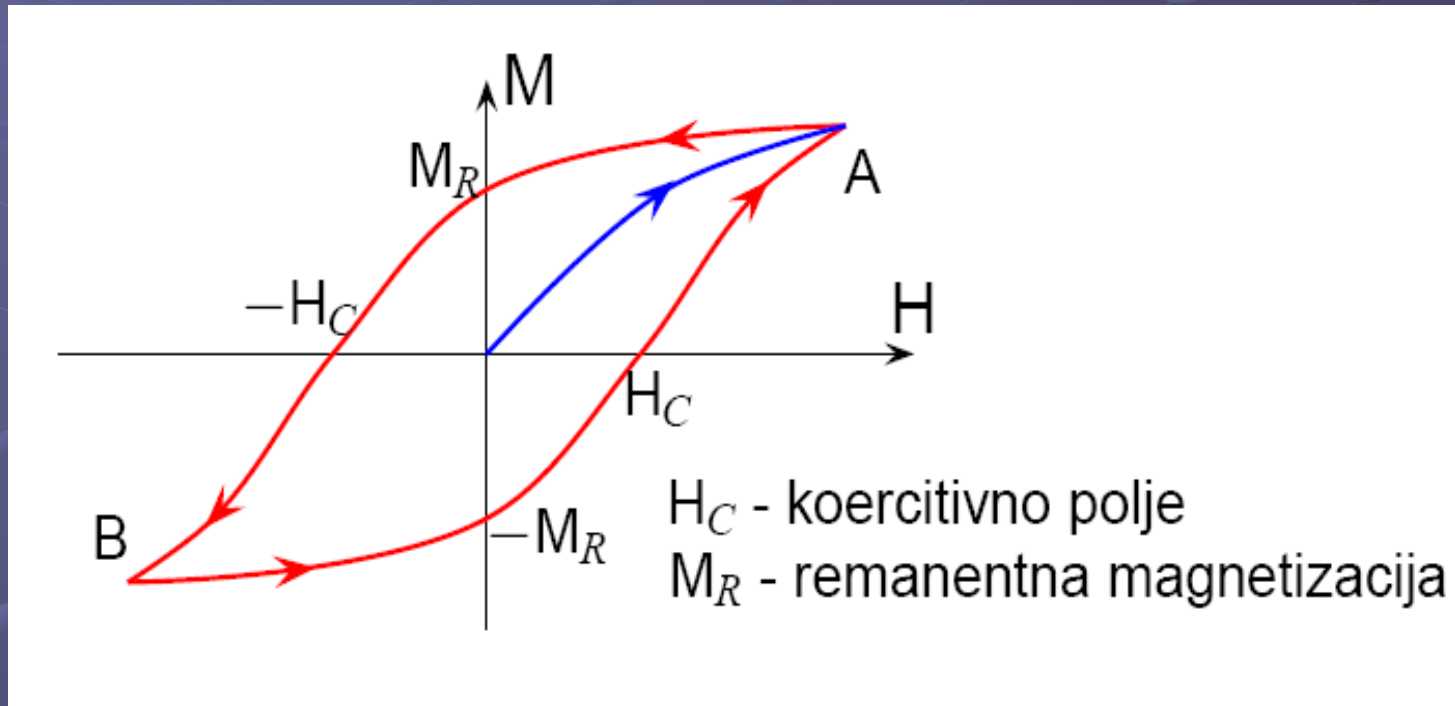
● $\vec{M} \neq 0$ i $\vec{H} = 0$ feromagnetizam

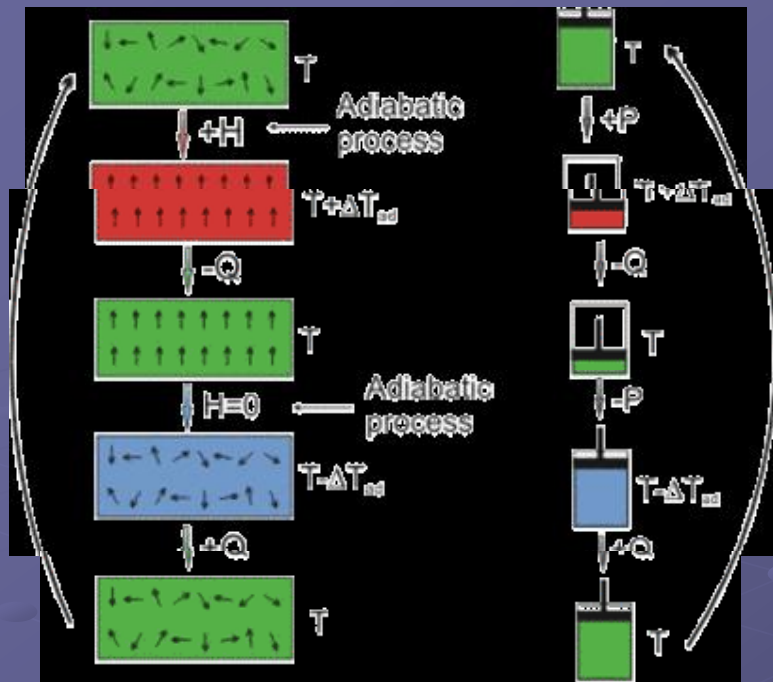
● $\langle \vec{M} \rangle = 0$, \vec{M} pravilno oscilira u prostoru
antiferomagnetizam

● $\langle \vec{M} \rangle \neq 0$, \vec{M} pravilno oscilira u prostoru
ferimagnetizam

Krivulja histereze

- Proces magnetiziranja makroskopskog uzorka





Magnetic refrigeration

Vapor cycle refrigeration

